# RAJASTHAN PUBLIC SERVICE COMMISSION, AJMER

# SYLLABUS FOR SCREENING TEST FOR THE POST OF SENIOR SCIENTIFIC OFFICER- NARCOTICS DIVISION (STATE FORENSIC SCIENCE LABORATORY, RAJASTHAN, JAIPUR)

### Unit I

Analytical Chemistry: Classification of analytical methods – Classical and Instrumental, volumetric, titrimetric and gravimetric techniques, selection of proper analytical techniques: types and range of determination, accuracy, precision and errors, sample preparation, handling of reagents with safety, density and viscosity measurements. Statistical Analysis: Mean, Mode, Median, Correlation and Regression analysis, Null Hypothesis, Variance, t-test, Chi-Square test. Type of Data, Measure of central tendency,

Dispersion of Data, Correlation, Probability and Proof.

#### **Unit II**

Analysis of unknown samples :-

Organic: Physical examination, element detection (N, S, Cl, Br, I, F), Functional Group analysis (-OH, -COOH, -NO<sub>2</sub>, -NH<sub>2</sub>, -CONH<sub>2</sub>, -CO-, -CHO, Hydrocarbons)

Inorganic: Qualitative analysis of cations and anions with special reference to cations i.e. As, Sb, Pb, Ba, Cu, Hg, Zn and Tl and anions i.e.  $NO_2^-$ ,  $NO_3^-$ ,  $S^{2-}$ ,  $SO_4^{2-}$ ,  $SO_3^{2-}$ , halides and cyanides.

Analysis of poisonous gases: CO, H<sub>2</sub>S, PH<sub>3</sub>, CH<sub>4</sub> and NH<sub>3</sub>.

## **Unit III**

Spectroscopic and other techniques:-

Unifying principles: Electromagnetic radiation, interaction of electromagnetic radiation with matter- absorption, emission, transmission, reflection, refraction, dispersion, polarization and scattering.

Basic principles, instrumentation and applications: UV- Visible, FTIR, AAS, Mass, Spectroscopy, Fluorescence and Phosphorescence spectrophotometry, ESR Spectroscopy. Fundamentals of Acids, Bases and Buffers, pH, p $K_a$ , and p $K_b$  values, principles, instrumentation and applications of pH metry, Potentiometry, Conductometry and Microscopic analysis in forensic Science.

#### **Unit IV**

Chromatography and Electrophoresis: General Principles and types of chromatographic techniques: Paper chromatography, column chromatography, Thin layer chromatography, adsorption chromatography, partition chromatography, Gas chromatography, Gas-liquid chromatography, Ion exchange chromatography, Exclusion (permeation) chromatography, affinity chromatography, HPLC, HPTLC, Capillary Chromatography and Electrophoresis.

#### Unit V

Basic Organic Chemistry: Important preparations and properties of alkanes, alkenes, alkynes, aromatic hydrocarbons, alcohols, phenols, carboxylic acids, aldehydes, ketones, amines and nitro compounds.

### Unit VI

Proteins: Classification, Structure and Properties, Molecular weight determination, Isoelectric point, coagulation and denaturation. Carbohydrates: Classification, Structure and Reactions. Fats and Lipids: Classification, Structure and Reactions. Alkaloids: Classification, Isolation and Identification.

### **Unit VII**

Extraction, isolation & identification of Alkaloids viz- Morphine, Codeine, Brucine, Strychnine, Nicotine, Atropine, Hyoscyamine, Cocaine, Heroin and Dhatura alkaloids. Extraction, isolation & identification of sedative, depressants, stimulants, opiates and drugs of abuse.

# **Unit VIII**

Medicinal Chemistry: General drugs, Designer Drugs, Drugs of abuse, mode of administration and pharmacological action of drugs of forensic importance. Drugs Act, Excise Act and NDPS Act.

## **Unit IX**

Narcotics drugs and psychotropic substances: Definition, types, appearance, production and chemical characteristics. Common terminology of various drugs. Drug action on central nervous system. Sampling and analytical techniques for qualitative & quantitative analysis.

### **Unit X**

Plants of Narcotic importance and their morphology: Papaver somniferum, Cannabis sativa, Coca plant and analysis of their active constituents.

Psychotropic substances: Amphetamines, Benzodiazepines and their derivatives. Barbiturates, Lysergides, Mascalines and Psilocybin etc.

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## **Pattern of Question Papers:**

1. Objective Type Paper

2. Maximum Marks: 100

3. Number of Questions: 100

4. Duration of Paper : Two Hours

5. All Questions carry equal marks

6. There will be Negative Marking

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